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3-1 Calculate the specific volume of an air-vapor mixture in cubic meters per kilogram of dry air when the following conditions prevail: t = 30 C, W = 0.015 kg/kg, and p t = 90 kPa. Solution: Equation 3-4. t s a a p p R T p R T - = = T = 30 C + 273 = 303 K R a = 287 J/kg.K P t = 90 kPa = 90,000 Pa Equation 3-2 t s s p p 0.622p W - = s 90 p 0.622p 0.015 - = 1.35 - 0.15p s

or CLICK HERE

Solution: q = mc p (t 2 - t 1) m = 2.5 kg/s c p = 1.0 kJ/kg.K t 2 = 30 C t 1 = -10 C CHAPTER 2 - THERMAL PRINCIPLES Page 2 of 5 Then, q = (2.5) (1.0) (30 + 10) q = 100 kw - - - Ans. 2-3.

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